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A global review on ambient Limestone-Precipitating Springs (LPS): Hydrogeological setting, ecology, and conservation

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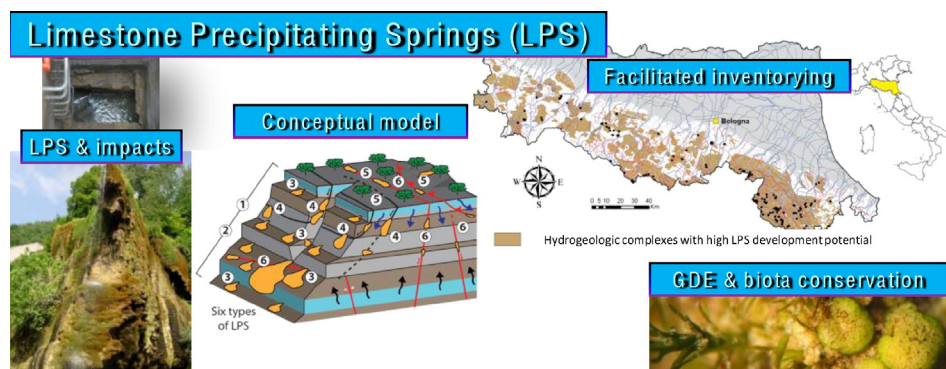
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HIGHLIGHTS

- Limestone Precipitating Springs (LPS) are ideal to study biocalcification.
- Spring-habitat protection is limited globally; in Europe there is a focus on LPS.
- We present a conceptual model to predict LPS occurrence to meet EU directives.
- Main impacts on LPS are water overdraft and lacking appreciation of their relevance.
- LPS should be a flagship to achieve widespread conservation of springs in general.

GRAPHICAL ABSTRACT



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ABSTRACT

Springs are biodiversity hotspots and unique habitats that are threatened, especially by water overdraft. Here we review knowledge on ambient-temperature (non-geothermal) freshwater springs that achieve sufficient oversaturation for CaCO_3 -by physical CO_2 degassing and activity of photoautotrophs- to deposit limestone, locally resulting in scenic carbonate structures: Limestone-Precipitating Springs (LPS). The most characteristic organisms in these springs are those that contribute to carbonate precipitation, e.g.: the mosses *Palustriella* and *Eucladium*, the crenophilous desmid *Oocardium stratum*, and cyanobacteria (e.g., *Rivularia*). These organisms appear to be sensitive to phosphorus pollution. Invertebrate diversity is modest, and highest in pools with an aquatic-terrestrial interface. Internationally, comprehensive legislation for spring protection is still relatively scarce. Where available, it covers all spring types. The situation in Europe is peculiar: the only widespread spring type included in the EU Habitat Directive is LPS, mainly because of landscape aesthetics. To support LPS inventoring and management to meet conservation-legislation requirements we developed a general conceptual model to predict where LPS are more likely to occur. The model is based on the pre-requisites for LPS: an aquifer lithology that enables build-up of high bicarbonate and Ca^{2+} to sustain CaCO_3 oversaturation after spring emergence, combined with intense groundwater percolation especially along structural discontinuities

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(e.g., fault zones, joints, schistosity), and a proper hydrogeological structure of the discharging area. We validated this model by means of the LPS information system for the Emilia-Romagna Region (northern Italy). The main threats to LPS are water diversion, nutrient enrichment, and lack of awareness by non-specialized persons and administrators. We discuss an emblematic case study to provide management suggestions. The present review is devoted to LPS but the output of intense ecological research in Central Europe during the past decades has clearly shown that effective conservation legislation should be urgently extended to comprise all types of spring habitats.

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1. Introduction: What are Limestone-Precipitating Springs (LPS)?

Springs may represent or feed valuable GDE (Groundwater Dependent Ecosystems; e.g., Kløve et al., 2011a, 2011b). Springs are unique habitats threatened by multiple impacts, either anthropogenic (water overdraft, e.g., Powell et al., 2015; contamination or drainage from subsurface excavations) or natural (e.g., the effects of climate change on the hydrologic cycle; Hartmann et al., 2014). One of the main reasons why springs are biodiversity hotspots (e.g., Cantonati et al., 2012a) is that they cover an extremely wide range of water chemistry, temperature, and environmental settings, and sustain a high within-habitat patchiness.

Spring types have been classified on the basis of diverse indicators (Alfaro and Wallace, 1994; Kresic, 2010): hydrology (discharge rate, e.g., Meinzer, 1923; Netopil, 1971; recession coefficient, e.g., Gargini et al., 2008; active vs. inactive, Fensham et al., 2015), hydrogeology (e.g., Civita, 1973), geology and morphology (e.g., Springer and Stevens, 2009), physico-chemistry (e.g., electrical conductivity, Total Dissolved Solids, e.g., Clarke, 1924), temperature (e.g., Glazier, 2009, with a transition between cool and thermal groundwater between 30 and 40 °C), flow conditions at the spring head (Thienemann, 1924), and biological characteristics, in particular vegetation (bryophytes + vascular plants; e.g., Ellenberg, 2009) and invertebrate communities (e.g., Gerecke and Di Sabatino, 1996; Schröder et al., 2006; Martin and Brunke, 2012). Furthermore, biota-based spring types can be recognized by using benthic algae forming colourings and/or structures identifiable with the naked eye (Cantonati et al., 2012b), or their diatom microflora (Cantonati et al., 2012c). A comprehensive multi-taxon classification by “Procrustes” analysis showed that each group of organisms provides useful specific spring characters that however are related to other groups only at the level of broad ecological categories (e.g., photoautotrophs, meiofauna etc.; Spitale et al., 2012). In the USA, twelve spring types were distinguished by geomorphology and recurring species (Springer and Stevens, 2009).

Virtually all groundwater is subject to some amount of geothermal heating but this may be very small or negligible in the case of shallow, subsurface, very short recharge–discharge systems. When significant geothermal groundwater heating can be excluded, spring water will be near the mean annual air temperature (MAAT) of the recharge area (cf. Pentecost, 2005), and it was recommended that these be renamed ‘ambient springs’ (e.g., Glazier, 2009).

There is no universally accepted temperature limit for spring water to be designated as thermal, i.e., water that was substantially heated by geothermal energy. In Europe, a temperature of 20 °C at spring emergence is generally accepted (e.g., Boch et al., 2005). Other limits to thermal water have been proposed such as, for instance, 36.7 °C (mean core temperature of humans; Pentecost et al., 2003) and 21.1 °C (70 °F).

Over most of the Earth, the mean annual air temperature near ground is <20 °C (see, e.g., the Köppen modified scheme of world climates). In some areas located in the subtropical high-pressure cells, and over most of tropical lowlands, however, mean annual air temperature is above 20 °C; in consequence, there, ambient-temperature springs can also be warmer than 20 °C (e.g., Carthew et al., 2006). Therefore, we avoid the terms ‘cool’ and ‘hot’ or ‘warm’ to designate spring-water temperature, but prefer to speak of ‘ambient’-temperature springs that are dealt with herein.

With respect to limestone deposition from springs, the temperature of the water at emergence was used for classification. Limestones deposited from thermal springs are commonly termed “travertine”, whereas limestones of ‘cool’ (non-thermal) springs are generally classified as tufa or calcareous tufa (e.g., Pentecost, 2005; Golubić et al., 2008). Both these terms, however, are problematic because they comprise an a priori interpretation of spring-water temperature (which may be difficult to deduce for fossil systems), and of the position of a given limestone within a spring depositional system (in fossil deposits, and even for parts of active spring-limestone deposystems, this may give rise to misinterpretations; see discussion in Sanders et al., 2011). The term tufa also cannot be reserved for highly porous (>10–15% porosity; see Ford and Pedley, 1996; Pedley, 2009) limestones from ‘cool’ springs, because similarly-high porosities are observed in actively-forming travertines of thermal springs. To avoid these confusions, we designate the deposits of Limestone-Precipitating Spring (LPS) neutrally as spring-associated limestones (SAL), irrespective of actual or interpreted water temperature, of derivation of spring waters (meteozone or thermogene), and irrespective of porosity (Sanders et al., 2011; Cantonati et al., 2012b, 2012c).

In general, composition and polymorphism of spring-related minerals are controlled by water chemistry rather than by the microbial communities mediating precipitation (Konhauser, 2007). Most SAL deposits consist of low-magnesian calcite (Table 1). This reflects the most widespread chemical composition of LPS, i.e., Ca^{2+} – HCO_3^- waters with smaller amounts of other common ions (mainly sulphate, chloride, Mg, Na, K, and dissolved silica). Ambient-temperature LPS with a Mg/Ca molar ratio ≥ 2.5 –3 are comparatively rare; these are characterized by precipitation of magnesian calcite and aragonite (Table 1). The precipitation of low-magnesian calcite is further impeded or modified by elevated concentrations of sulphate, orthophosphate, and some groups of organic substances (e.g., Bischoff and Fyfe, 1968; House, 1987; Plant and House, 2002; Lin et al., 2005; Fernández-Díaz et al., 2010). The impact of any of these compounds on crystal growth, however, seems to depend on many factors (e.g., concentration, pH, association with other ions or molecules), and details are far from resolved. In dysoxic to anoxic LPS, if present even in very low concentrations, Fe^{2+} completely blocks CaCO_3 precipitation (cf. Dromgoole and Walter, 1990). At spring emergence, thus, first the Fe^{2+} has to be removed by iron-bacterial oxidation (e.g., *Gallionella*, *Leptothrix*) into virtually-insoluble Fe^{3+} –hydroxides (e.g., Søgaard et al., 2001; Chan et al., 2009). Only when the Fe^{2+} is exhausted, farther downstream, CaCO_3 precipitation can start, resulting in ‘mineralogically-zoned’ iron oxide/ CaCO_3 deposits (Sanders et al., 2011).

The crystal habit of pristine LMC precipitates (pristine = crystal precipitated from and still bathed in its parent solution) is extremely variable and ranges, for instance, from perfect ditrigonal scalenohedra to crystal skeletons to spheroids to needles, to name a few; similarly, crystal size ranges from nanometer- to millimeter scale (see, e.g., Freyret and Verrecchia, 1998; Pentecost, 2005; Turner and Jones, 2005; Shiraiishi et al., 2008). Whereas some correlation of crystal habit with water chemistry is obvious (see above), further influences most probably are degree of oversaturation and turbulence at microhabitat, rate of nucleation of crystals or subcrystals, fluctuations of water chemistry or water supply and, finally, biological mediation of precipitation.

Table 1
Mineralogy and polymorphy of most common CaCO₃ minerals formed from ambient-temperature springs.

Mineral; composition; crystal system	Chemical controls	Geological controls	Remarks
Low-magnesian calcite; CaCO ₃ with <4 mol% MgCO ₃ ; trigonal	Precipitates from Ca ²⁺ –HCO ₃ [−] waters low in Mg ²⁺ (common for LPS)	Rocks and/or sediments rich in Ca; source of HCO ₃ [−] : dissolution of rock or sediment, soil cover, hypogenic CO ₂ ascend	Most common mineral from LPS. Takes many crystal habits from rhombs to fibers
Magnesian calcite; CaCO ₃ with >4 mol% MgCO ₃ ; trigonal	Forms from waters with Mg/Ca molar ratio above c. 2.5–3 (rare for LPS)	As above, but with Mg source, e.g., mafic-ultramafic rocks, magnesite deposits, or Mg-rich gangue of deposits	Rare mineral from ambient-temperature LPS. Typically in fibrous crystals that form fringes and botryoids
Aragonite; CaCO ₃ with <4 mol% MgCO ₃ ; orthorhombic polymorph of CaCO ₃	Forms from waters with Mg/Ca molar ratio above c. 2.5–3, and/or from waters with high HCO ₃ [−] concentration (rare for LPS)	As above for magnesian calcite	Rare mineral from ambient-temperature LPS. Common in hot-spring limestones (not treated herein). Typically in fibrous crystals forming fringes and botryoids

Besides thermal/geochemical/mineralogical characteristics, other aspects of LPS are locally important around the world, e.g. the cultural, environmental, and societal significance of LPS. For instance, in Germany, a small LPS, deposited a large SAL over several thousands of years, known as the largest ‘channel on a stone ridge’ (Ger. *Steinerne Rinne*) in Germany (40 m long and 5 m high). The ‘Growing Rock’ (Ger. *Wachsender Stein*) of Usterling, also known as Johannes Rock after John the Baptist, is a natural monument in Usterling (Landau, Bavaria). Its oldest representation can be found on a late Gothic altar in the village church of St. John of Usterling: In one image, Christ's baptism by John is relocated to this growing rock – a cultural-history curiosity. The illustration shows the natural monument as it should have looked in the 1500s. In 2006 it was admitted to the list of excellent National Geotopes of Germany. Over the rock stands St. John's Chapel, and at its foot a chapel shrine with a wooden Johannes figure (Bauer et al., 2009). In Section 6 we discuss and even more striking example in Italy (Labante), where a scenic LPS is still active in spite of multiple impacts, it includes impressive fossil SAL extensively used since antiquity (Etruscans), and it is an emblematic example of the development of environmental policies on LPS in the last decades.

This review paper summarizes knowledge on freshwater, ambient-temperature LPS and their biota, on LPS distribution and conservation status worldwide, and on major impact types (e.g., water overdraft, P enrichment, lack of awareness) while providing management suggestions. At the same time the review paper presents a novel conceptual model to predict LPS occurrence from environmental settings, validated by its application to worldwide case studies (own + literature) and discussed illustrating a regional application. One emblematic case study will exemplify the main impact types.

2. Biota and biocalcification

Characteristic and common taxa of the flora and fauna found in LPS are summarized in Table 2, with indication of the microhabitat(s) they typically occupy within LPS systems, and shown in Fig. 1. Whereas elements of the flora can contribute to shape limestone deposition in springs, elements of the fauna of LPS are frequently hindered in their life functions by limestone deposition.

2.1. Flora

Besides the preponderant role of physical CO₂ degassing by turbulence, pressure release, and temperature increase of waters from the spring origin to downstream that shifts water chemistry towards

oversaturation for CaCO₃ (Merz-Preiß and Riding, 1999; Chen et al., 2004), active biogenic processes can lead to deposition of low-magnesian calcite by photosynthetic withdrawal of HCO₃[−] and CO₂ (Schagerl and Wukovits, 2014). In addition, passive trapping and binding of particles and enrichment of HCO₃[−] and Ca²⁺ on organic surfaces (polysaccharides or proteins) can take place (e.g., Merz-Preiß and Riding, 1999; Kawaguchi and Decho, 2002; Turner and Jones, 2005; Ditttrich and Sibling, 2010). Downstream changes in oversaturation related to photosynthesis within spring streams may amount to a few percent only (Pentecost, 1992, 2005; Shiraishi et al., 2008), except for situations where *Rivularia* dominates and/or in larger spring-fed streams (e.g., Rott et al., 2000) where no inorganic limestone precipitation takes place (Shiraishi et al., 2008).

Several phylogenetic lines among oxygenic photoautotrophs are supposed to contribute to biocalcification processes in LPS, including cyanobacteria, eucaryotic algae, and bryophytes (Fig. 1a–i).

Bryophytes (Fig. 1a–e) are the most characteristic organisms that contribute probably to a larger extent to carbonate precipitation in springs, and to the formation of pools and cascades by CO₂ consumption in photosynthesis; this holds in particular for widespread (circumpolar and warm temperate) mosses, such as *Eucladium verticillatum* and *Palustriella* spp. (*Cratoneurion commutati* in Europe, Dierssen, 2001; Tomaselli et al., 2011; *Palustriella falcata* and *Cratoneurion filicinum* in America, Flora of North America Editorial Committee eds., 1993 +).

A specific eukaryotic alga potentially confined to LPS is the desmid (green alga) *Oocardium stratum* (Fig. 1a–b) Sanders and Rott, 2009; Rott et al., 2012; Linhart and Schagerl, 2015), which shows an almost worldwide distribution (at least circumpolar, temperate, and tropical); it is documented for China, Cuba, India, North America, and a few locations all over Europe (see summary in Linhart and Schagerl, 2015).

Cyanobacteria are found widespread in both freshwater and marine carbonate depositional environments often related to aquatic in-transition-to terrestrial habitats (stones, soils). *Rivularia* is one of the most interesting in relation to LPS (Fig. 1g–i). This genus is represented in freshwater and marine habitats (Freytet and Verrecchia, 1998).

Diatoms (Fig. 1f) are seasonally abundant in LPS, mainly during the cold and low-illuminated season (e.g., Sanders and Rott, 2009; Linhart and Schagerl, 2015). *Gomphonema calcareum*, in spite of its specific epithet, is usually not observed to form carbonate-encrusted colonies (Levkov et al., 2016). Other *Gomphonema* species are much more common in LPS (cf. Table 2), in particular *Gomphonema lateripunctatum*, an indicator species of the spring type ‘carbonate hydropetric springs (= rock-face seepages) and LPS’ (Cantonati et al., 2012c). Macroscopic colonies of the hard-water diatom species *Cymbella excisiformis* were

Table 2
Biota involved in the formation of LPS and dependent or associated with LPS ecological niches.

Taxon	LPS formation	LPS-associated ecological niches
PHOTOAUTOTROPHS		
Cyanobacteria		
<i>Rivularia</i> spp.	x	
<i>Homoethrix crustacea</i>	x	
<i>Phormidium incrustatum</i>	x	
<i>Tolypothrix</i> sp.	x	
<i>Scytonema</i> (<i>Myochrotes</i>) spp.		Hy
Diatoms		
<i>Achnanthes trinodis</i>	Eu-LPS	Hy
<i>Brachysira calcicola</i>	Eu-LPS	
<i>Gomphonema lateripunctatum</i>		Hy
<i>Fragilaria distans</i>	Eu-LPS	
<i>Delicata minuta</i>	Eu-LPS	Hy
<i>Denticula elegans</i>	Eu-LPS	Hy
<i>Cymbella diminuta</i>	Eu-LPS	
<i>Cymbella excisiformis</i> (macroscopic colonies)		Ls
Desmidiacean green algae		
<i>Oocardium stratum</i>	x	
Yellow-green algae		
<i>Vaucheria</i> sp.	x	
Mosses		
<i>Palustriella commutata</i>	x	
<i>Eucladium verticillatum</i>	x	
<i>Cratoneurion filicinum</i>	x	
Vascular plants (Brassicaceae)		
<i>Cochlearia bavarica</i>		Br
INVERTEBRATES		
Caddisflies		
<i>Rhyacophila pubescens</i>		Ls
Moth flies (Diptera Psychodidae)		
<i>Pericoma trifasciata</i> gr.		Eu-LPS, Hy
Midges (Diptera Chironomidae)		
<i>Tanytarsus emarginatus</i>		Ls
<i>Rheotanytarsus reissi</i>		Ls
Gammarids (Crustacea, Amphipoda, Gammaridae)		
<i>Gammarus fossarum</i>		Eu-LPS, Lc, Ls
Salamanders (Caudata, Salamandridae)		
<i>Salamandra salamandra</i>		Lc

Eu-LPS = Eucrenal of LPS, Lc = Limnocrenic, Hy = Hygropetric, Ls = Lotic sectors, Br = Bryophytes & fine-grained sediments.

observed in a LPS, however not encrusted with calcium carbonate (MC unpublished data).

Bryophytes may calcify directly on their surface so that dead organic parts (e.g., stems) remain enclosed in the calcites and degrade gradually (Fig. 1e); in many cases, additional calcification of organisms living on the bryophyte plants, such as diatoms or *Oocardium stratum* plays a considerable role providing a large surface for smaller biota. Eucaryotic microalgae, such as *Oocardium stratum*, can precipitate low-magnesian calcite at vertical accumulation rates of up to 10 mm y⁻¹ (Sanders and Rott, 2009); this alga seems to grow only in waters conducive to low-magnesian calcite precipitation (Rott et al., 2012). The initial calcification of *O. stratum* is highly variable with respect to basal crystallite fabrics, so that several (at least three major within a cross section of the Alps and within the spring variability) of most probably biotic induced calcification types (calcites) were related to environmental variations in space and time of microhabitat conditions (e.g., pH, temperature, CO₂ oversaturation; Rott et al., 2012).

Several cyanobacterial taxa show indeterminate calcification fabrics found also in LPS, although the precise mechanisms of cyanobacterial

calcification are not yet clarified in detail (see e.g., Merz, 1992; Freydet and Verrecchia, 1998; Shiraishi et al., 2008). An interesting example of specific cyanobacterial calcites is known from the genus *Rivularia* (Fig. 1h-i) with a specific carbonate microfabric identifiable as of cyanobacterial origin (e.g., Flügel, 2004). *Rivularia* often forms primarily concentric layers within their young hemispheric colonies with calcite rhombs of micro- to orthospar size embedded in the mucilage layers between filament sheaths until finally the trichomes are completely fixed in calcite spar (Obenlünenschloss, 1991). Whereas early calcification is influenced by oversaturation for CaCO₃, along spring streams other factors such as local current velocity and turbulence similarly are important (e.g., Freydet and Verrecchia, 1998; Sanders and Rott, 2009; Gradzinski, 2010).

Diatoms commonly are considered as non-influential with respect to spring calcification (Fig. 1f), yet they appear to contribute by inducing the formation mainly of loose, micritic to sparitic calcium carbonate sediment not firmly bound into specific calcification fabrics (Sanders and Rott, 2009). Detailed microscopic and SEM studies show that diatom frustules associated with CaCO₃ are widespread. In the mucilage of diatom mats, calcite crystals of micro- to orthospar size are common. In addition, diatom frustules and also their stalks are frequently embedded in calcite crystals, and loose aggregates of frustules embedded in clumps of micrite to microsparite are common (Wallner, 1935; Sanders and Rott, 2009). Whether the diatom frustules and their stalks are just passively trapped within the crystals or crystal clusters, or to what extent they took an active role in inducing CaCO₃ precipitation, however, is unknown. In mid-latitudes with a distinct climatic seasonality, the seasonal changes of spring biota associated with changes of water chemistry, temperature and illumination typically impart annual or seasonal laminations to SAL deposits (e.g., Kano et al., 2003; Kawai et al., 2006; Shiraishi et al., 2008; Sanders and Rott, 2009; Arenas et al., 2010).

2.2. Fauna

LPS typically show low faunal diversity. Most animals adapted to this habitat live in pool areas below cascades or in the aquatic-terrestrial interface where allochthonous organic matter accumulates (Table 2). The proportion of specialized spring-dwellers is low, and the fauna of LPS is very poor in character species (Zöllhöfer, 1997; Martin and Wischniowski, 2014). In general, faunal richness in springs is found to be directly influenced by microhabitat diversity, in particular by the availability of stable transition-habitats in the aquatic-terrestrial ecotone. Since the limestone cover produced by CaCO₃ deposition imparts homogenization of microstructures and reduction of microhabitat richness, a low animal diversity is typical for LPS (Dürrenfeldt, 1978; Martin and Wischniowski, 2014).

Animal species able to persist in the petrified channels are mostly rithobiotic generalists which migrate from their preferred habitat (mountain streams) into spring brooks (in lotic sectors: gammarids, in pools: salamander larvae and some taxa of Trichoptera).

An exception is the case-less larva of the caddisfly *Rhyacophila pubescens*, which can be considered an outstanding element of LPS in Europe. It is strictly bound to springs and low order stream sectors rich in calcium-carbonate deposits. A genetic analysis showed a rather high isolation of local populations, probably due to habitat fragmentation combined with a low dispersal rate of adults (Engelhardt et al., 2011). With regard to other genera and families, the trichopteran fauna, usually very diversified in springs, is uniform and poor in LPS. Kühn (1940) reports an extreme example of case-building caddisflies literally “caught” and immobilized by quickly growing calcium-carbonate crusts. Most probably, the involved specimens were at a late nymphal stage, at the beginning of pupal quiescence.

Two phenomena observed among Diptera larvae may be mentioned in this context: At oversaturation for CaCO₃, larvae of *Pericoma* species (Psychodidae) tend to become encrusted by calcium carbonate

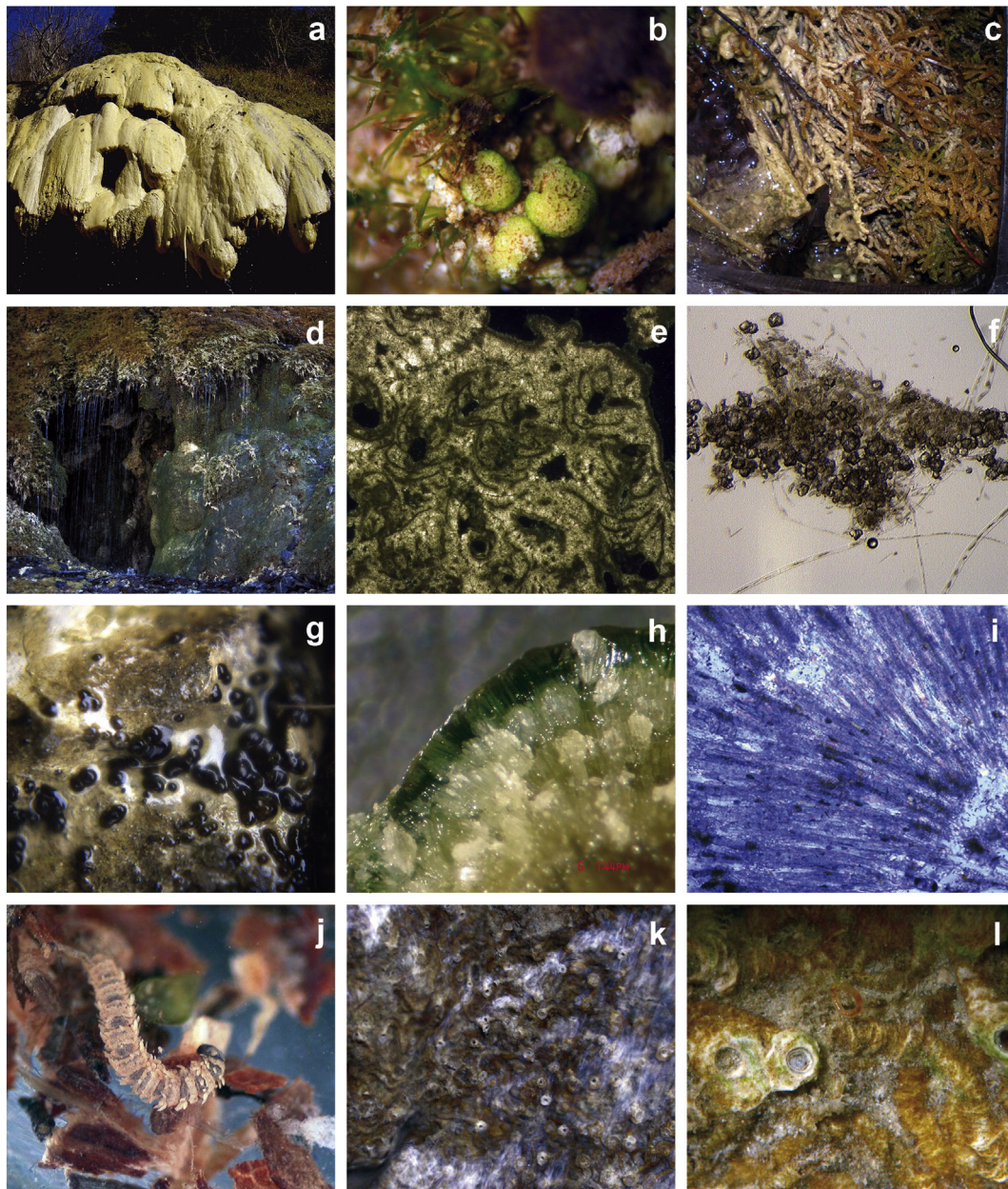


Fig. 1. LPS biota and calcification products. (a) Waterfall tufa curtain composed mainly of *Oocardium* calcite (yellow area) (Austria, Northern Alps); width of view in middle ground approximately 6 m. (b) Ensemble of moss (*Palustriella commutata*) and of calcifying hemispherical aggregates of the desmidiacean microalga *Oocardium stratum* (Austria, Northern Alps); width of view approximately 3 cm. (c) The moss *Palustriella commutata* from a LPS (some plantlets are completely covered by carbonate precipitates). (d–e) Bryophyte tufa: (d) Active bryophyte tufa; in many cases the bryophyte tufa is percolated by calcifying waters that may intermittently emerge to surface along morphological knicklines; Egg, Austria; width of view ca. 1 m. (e) Fossil bryophyte tufa; the bryophyte leaves are preserved as delicate structures of micrite (grey); the former bryophyte stems are preserved as open pores (black); the space between the mosses is filled with fringes of fibrous aragonite; Tobadill, Austria; crossed nicols; width of view 4.2 mm. (f) Flake, removed from sticky diatom mat, showing rhombohedral crystals and crystal aggregates of calcite floating within the mat's mucilage; width of view 1.4 mm. (g–i) *Rivularia* sp. (cyanobacterium): (g) Colonies (a few mm in diameter) on rocky material, width of view: 3 cm. (h) Section across a colony calcified in its lower part (light-grey area), blue-green layer < 1 mm. (i) Thin section in epoxy of a *Rivularia* limestone from the River Alz (outlet of Chiemsee, Bavaria, southern Germany); diameter of filaments is c. 10 μm . (j) Carbonate-encrusted *Pericoma* sp. larva (drain fly; Diptera, Psychodidae); length of the specimen c. 8 mm. (k–l) Tufa including many midge-larvae cases (Diptera, Chironomidae) in LPS-fed streams (Ger. *Chironomidentuff*); width of view ca. 5 cm (k), 1 cm (l) (Photos k–l: T. Ekrem). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

(Feuerborn, 1923; Vaillant and Withers, 1993). The CaCO_3 precipitates preferably around the long dorsal setae which take the appearance of horn-like appendages, but also on other integument structures (Fig. 1j). The animals are obviously not hampered in their lifestyle, and even have advantages such as mechanical body protection and camouflage. Larvae of the non-biting midges (Chironomidae) *Tanytarsus emarginatus* and *Rheotanytarsus reissi* normally live in self-produced silk tubes covered by fine sand. In springs and streams with rapid CaCO_3 precipitation they form dense populations associated

with *Fissidens* spp. moss and cyanobacteria of the genus *Rivularia* in biohermal structures. This so-called “chironomid tufa” (Fig. 1k–l) grows passively without active contribution to CaCO_3 precipitation by the dwelling animals. The abundance of such populations, however, suggests that they profit from this lifestyle (Thienemann, 1934; Kühn, 1940; Burmeister and Reiss, 2003).

As a general rule, decreased faunal diversity correlates with rate of carbonate precipitation in LPS. In “moderate-rate LPS”, formation of riparian transition zones with organic debris and moss/macrophyte

vegetation may allow for the formation of pools with a typical crenobiontic fauna. In Central Europe, the water mite *Lebertia helocrenica* is a potential character species of such habitats. The species, however, is known from three sites only (in the Southern and Northern Alps and Prealps; Gerecke, 2009); this highlights the need for more faunistic research dedicated to “moderate-rate LPS”.

3. Worldwide conservation status of Limestone-Precipitating Springs

In Europe, the conservation legislation of LPS is based on the Annex I of the Habitat Directive (EU-HD, 1992) in which LPS are listed as “Petrifying springs with tufa formation (*Cratoneurion*)” (EU-Code 7220). Among spring habitats mentioned in the Habitat Directive, LPS are by far the most widespread. However, for the sake of completeness, we note that the Habitat Directive Annex I includes also two very special and geographically-localized types of springs: “Inland salt meadows” (*Puccinellietalia distantis*, EU-Code 1340, including some inland saline springs), and “Fennoscandian mineral-rich springs and spring fens” (EU-Code 7160).

Inclusion in the Habitat Directive allows designated LPS to be preserved in the frame of the Natura 2000 coordinated network of protected areas. Furthermore, some LPS sites are protected as National Parks or UNESCO World Heritage sites, such as the Plitvice Lakes National Park in Croatia.

In connection to LPS and spring protection, it might be also worth mentioning that downstream and in close proximity to SAL, “springs with iron precipitates” can occasionally be found in form of ‘mineralogically-zoned’ iron oxide/calcium-carbonate depositing streams. This relatively-rare type of spring requires anoxic to dysoxic groundwater conditions, and is remarkable from a mineralogical and biological point of view (cf. Sanders et al., 2011).

On the basis of an international consultation from all continents (see Acknowledgements), it can be stated that the situation in Europe (i.e. spring protection focused on LPS) is very peculiar, and cannot be found anywhere else in the world. Internationally, comprehensive legislation for the protection of spring habitats in general is still relatively rare. In countries where it is available, it covers all spring types, or broad categories of spring habitats (e.g., two categories are considered in Australia: Tertiary springs fed by localized aquifers, and Discharge springs fed by the Great Artesian Basin; Renee Rossini, University of Queensland, St. Lucia, Australia, personal comm.).

However, there are some interesting local situations that combine approaches, or partly differ from the situation as described above.

In Finland, there is no special legislation for LPS but the EU Habitat Directive applies. Interestingly, in addition, at the national level, Finland has the ‘Water Act’ which protects all pristine or close-to-pristine springs as habitats from damaging, and the ‘Forest Act’ which protects the forested marginal surroundings of all pristine/close-to-pristine springs from excess forestry (Jari Ilmonen, Biodiversity Research Programme, Finnish Environment Institute, Helsinki, Finland, personal comm.).

The USA have no protection for LPS (Larry Stevens, MNA Springs Stewardship Institute, Flagstaff, AZ, personal comm.) but there is a very peculiar situation in the State of Minnesota that has undertaken a specific conservation step in favour of LPS.

In Minnesota, there is a state law and associated rules (Minnesota Statutes 103G.223 Calcareous Fens; Minnesota Rules Chapter 8420.0935 Standards and Criteria for Identification, Protection, and Management of Calcareous Fens) aimed specifically at protecting what are defined as calcareous fens, which are peat-accumulating wetland areas supported by upwelling calcium/magnesium-carbonate rich groundwater, often having carbonate precipitates (tufa or marl). This law does not apply to other types of springs, nor are there any other regulations specifically focused on springs (Doug Norris, Wetlands Program Coordinator,

Minnesota Dept. of Natural Resources, Division of Ecological and Water Resources, St. Paul, MN, USA, personal comm.).

At worldwide scale water has since long been treated as an economic good. The “sanitary revolution” of the 19th century saw the demand for public ownership and management. This determined an emphasis on the public-good nature of water and led to the development of strongly-subsidized public systems. In the late 1980s, however, there was diffuse “privatization” of public services with all related problems of setting tariffs and prices (Rogers et al., 2002). As a consequence, springs are even more threatened since disregarded as natural habitats, with a strong tendency to consider them exclusively as a source of primarily (economically) precious water resources.

As other freshwater environments, LPS are also threatened by nutrient enrichment. Recent findings of *Oocardium stratum* in 8 springs and spring complexes within a N-S transect across the Alps were mostly from sites with low TP ($<10 \mu\text{g L}^{-1}$) but variable nitrate concentrations ($>2000 \mu\text{g L}^{-1}$ in 3 out of 5 sites). This could be an indication that this bioindicator is impeded by excess phosphates (House, 1987; Rott et al., 2012).

4. A conceptual model on ambient-temperature Limestone-Precipitating Springs distribution

Based on direct observations and revision of available published and unpublished datasets, we here present a conceptual model on the formation of LPS, with a focus on geological structure. The main aim of this model is to highlight how the structural-stratigraphic framework controls the hydrologic system in creating compartments, baffles, barriers and preferential groundwater flow pathways. In this framework, the proposed classification specifically provides insights on the subsurface structural architecture and physiography of the geological setting, which have been often overlooked in the concerning literature so far, and thus intended to integrate and complement the other widely used classifications based on different approaches considering the surface expressions of such deposits (see e.g. Pedley, 1990; Ford and Pedley, 1996; Pentecost, 2005; Pedley, 2009; Jones and Renaut, 2010).

Our database also includes those comparatively rare situations in which carbon dioxide is not derived from the atmosphere and/or the soil cover (e.g., Celico et al., 2010), but from deep sources related to hypogenic processes such as anaerobic degradation of hydrocarbons (e.g., Chakraborty and Coates, 2004) and metamorphism of carbonate rocks (e.g., Glassley, 1983; Ague, 2000; Bissig et al., 2006). We however exclude those cases in which “warm” water (compared to the environmental temperature; see above) is involved, and in which CO_2 derivation is directly related to magmatic and hydrothermal processes.

4.1. LPS structural background: types and significance

The model proposed focuses on structural heterogeneities and the related distribution of groundwater flow paths, assuming favourable biological, physical, and chemical conditions for LPS formation at the surface, already described in the classically used schemes deposits (see e.g. Pedley, 1990; Ford and Pedley, 1996; Pentecost, 2005; Pedley, 2009; Jones and Renaut, 2010). The model encompasses six (6) LPS types (Fig. 2) representing different associations of interacting and overlapping processes and products.

Type 1 LPS develops from well to partly consolidated lithologies, where the aquifer is defined by primary (e.g., depositional) and secondary (e.g., diagenetic) matrix permeability due to porosity (e.g., Becker and Shapiro, 2000). The groundwater flow is funnelled and/or vertically compartmentalized. Eventual emergence as an LPS is mainly controlled by depositional-diagenetic discontinuities (e.g., stratigraphic surfaces, diagenetic fronts; Mayo et al., 2003).

Type 2 LPS encompasses situations where the contribution from the matrix is negligible, and the overall bulk permeability of the hydrologic

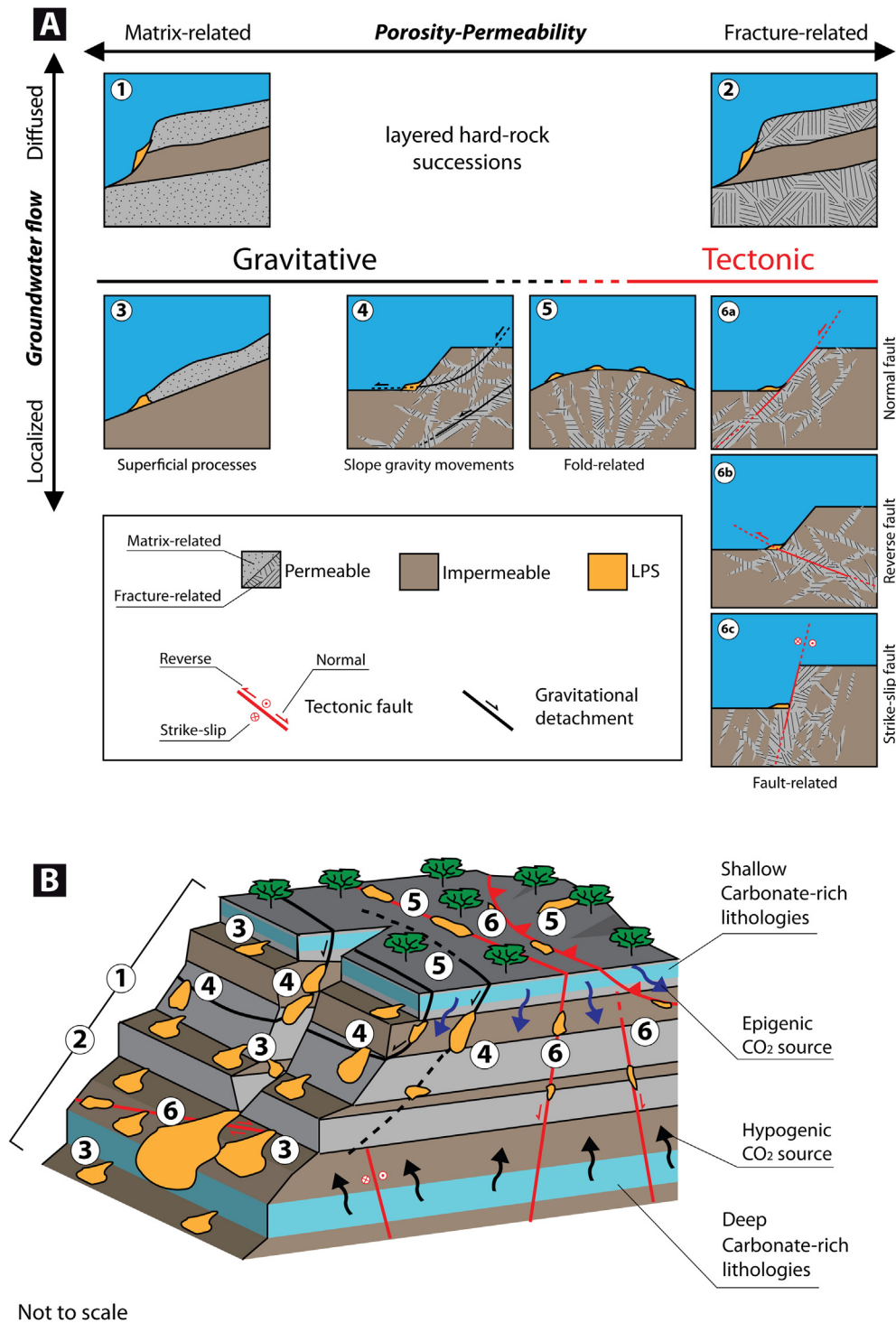


Fig. 2. Diagram illustrating the conceptual model. A. Schematic 2D representation of the 6 type-cases. B. Summary of the occurrence of the type-cases in an hypothetical 3D environment.

system is provided by fracturing (Motyka, 1998). In carbonate rocks, bulk permeability can be enhanced by meteoric dissolution, eventually leading to karst systems; these are not specifically discussed herein.

Type 3 LPS refers to Earth-surface processes resulting in unlithified to partly lithified, mostly coarse-grained sediment bodies with different fabrics, such as talus, alluvial fans, glacial deposits, as well as rock slides and rock avalanches. LPS are concentrated at the margins of the highly porous and permeable sediment bodies, or where the groundwater level intersects the surface of a thicker deposit (e.g., rockslide masses) (e.g., Sanders et al., 2011). Locally, LPS discharge to the surface is

associated with significant cementation within the permeated sediment bodies.

Type 4 LPS is associated with shallow- to deep-seated gravitational slope deformations. This type of LPS is typical for mountainous areas with high-gradient streams and steep relief. In the Apennines, slope gravity movements of km-scale slabs (mainly calciturbidite successions and ophiolites) within shale-rich chaotic complexes act as perched isolated aquifers that are effective in producing LPS (Chelli et al., 2013; Gargini et al., 2014; Carlini et al., 2015; Segadelli et al., submitted).

Type 5 LPS indicates situations where LPS development is due to fracture corridors arrays related to folding and bending of the hydrogeological formations. The classical recurrent example comprises a topographic-scale antiform breached by upward-fanning, along- and cross-fold fracture sets, representing preferential fluid escape pathways to the surface (e.g., Evans and Fischer, 2012; Ogata et al., 2014). It is important to stress that in this particular case the upward migration of cool water is not driven by temperature but overpressure, due to hypogenic (see below) CO₂ degassing in shallow, compartmentalized aquifers (e.g., Dockrill and Shipton, 2010).

Type 6 LPS comprises aquifers partitioned into highly permeable fault damage zones and low permeability fault cores (e.g., Bense et al., 2013). In these cases, LPS occurrence is related to fault pattern. The sub-groups 6a, 6b, and 6c are differentiated according to fault character: normal, reverse, or strike-slip. The permeability of fault zones is controlled by the degree of cementation of fracture porosity, and therefore related to depth of deformation, strain type and rate. Along with the fracture network sustaining effective rock leaching in the subsurface, associated morphological relief is an additional factor in providing a structurally controlled pressure gradient for meteoric water percolation.

Among the cases in which the fracture-related contribution to permeability prevails, there is a gradual transition from purely gravitational (Type 4 LPS) to entirely tectonic (Type 6 LPS) processes (see Fig. 1a). Deep-reaching fault systems (Type 6) and large-scale fracture corridors' arrays (Type 5) may promote ascend of hypogenic CO₂-rich waters, not necessarily hot, and unrelated to magmatic/hydrothermal activity. In these cases, the CO₂ can be provided by i) oxidation of thermogenic methane, and ii) metamorphism of deeply buried carbonate rocks (e.g., Frey et al., 2015).

The concepts introduced above are similar to those commonly used in the oil and gas industry for the characterization of dual porosity-permeability reservoir systems (e.g., Spence et al., 2014), here adapted to conditions where groundwater is the permeating fluid.

4.2. Worldwide occurrence of LPS types

Through a careful and detailed review of the inherent literature we compiled a database of recognized LPS types and their locations worldwide. As already pointed out above, warm-water-related LPS strictly associated to hydrothermal and magmatic/volcanic processes are excluded. The entire data collection is represented and summarized in Table 3 and Fig. 3.

This compilation, which is function of data availability/quality and intensity of studies in the related regions, is not intended as an exhaustive database for LPS occurrence, but a geographical distribution of examples used to validate the model. The importance of a robust global database, which is beyond the scope of this work, and the general guidelines to a shared workflow are pointed out in the next section.

Nonetheless a general overlap with the distribution of carbonate rock outcrops exists, LPS appear unrelated to specific environments or lithologies, being by far apparently and relatively underrepresented in karst systems. This suggests that favourable physical-chemical boundary conditions to LPS development might be achieved by different means and interactions, and that the structural control exerted by the geological framework appear to play a fundamental role.

5. Model application: Limestone-Precipitating Springs territorial information system of the Emilia-Romagna Region (Italy)

Extended inventories on the LPS distribution are often missing or incomplete, mostly because thoroughly field searches of entire geographic areas would be extremely expensive and time-/energy-consuming. Information on their occurrence and conservation status is thus absent or poor.

To overcome this limitation, since 2010, the Geological, Seismic and Soil Survey of Emilia-Romagna has activated:

- 1) the inventorying and cataloguing program of exploited springs (<http://ambiente.regione.emilia-romagna.it/geologia/cartografia/webgis-banchedati/sorgenti-unita-geologiche-sede-acquiferi-appennino>. Italian version only);
- 2) the mapping of the main host hydrogeological complexes;
- 3) the "Habitat map" cartography within Natura 2000 network and regional Parks (<http://ambiente.regione.emilia-romagna.it/parchi-natura2000/consultazione/cartografia-interattiva>. Italian version only).

The typical aquifers of the Northern Apennines are siliciclastic-calcareous turbidites and ophiolites, and the major perennial springs suffer the impacts of a variety of anthropic activities (e.g., mining, drinking water withdrawal, as for the Labante spring, see below). Few major springs contribute also to the maintenance of environmental flows in streams of the higher mountain range. Another significant threat to these important groundwater resources and related biota is the forced drainage induced by tunnelling for transport infrastructures and/or water-abstraction purposes; relevant are the impacts induced by the drilling of the High Speed Railway connection between Bologna and Florence (Gargini et al., 2008; Vincenzi et al., 2014) that caused the permanent vanishing of natural environmental flows during the dry season (summer) and the complete desiccation of major springs.

The inventorying program of the Emilia-Romagna Region pointed out the existence of 185 LPS (Fig. 4). These initiatives are promoted to abide by the requests of the Groundwater directive (GWD-European Commission, 2006).

Comparing the location of LPS with the distribution of the main hydrogeological complexes in the Emilia Romagna Region, it can be noticed that they preferentially occur near perennial springs fed by groundwater circulation inside carbonate dominated rock aquifers.

The map shown in Fig. 4 represents a first evaluation of the potential LPS-prone areas. In particular, from a first screening, the LPS appear to occur mainly within or along the perimeter of the major hydrogeological complexes. Such areas are the starting targets for dedicated investigation strategies aimed to evaluate the true potential for LPS occurrence.

Given the importance of these habitats, any decision-making policy aimed at protecting and managing LPS requires the prompt availability and consultation of geographic information stored in a comprehensive database accessible to the public. This kind of Territorial Information System represents the most simple and suitable mean to accomplish this task.

6. Case study and suggestions for management

We provide suggestions for an effective and sustainable management discussing a case study in which LPS are left in poor condition because of marked water diversion (total during long drought periods), and because of limited awareness of the conservation value of these fragile environments.

The Labante spring (Fig. 5; Table 4) is a LPS located in the northern Apennines (Castel d'Aiano, Bologna). The spring is an important source of drinking water and a site of high environmental and touristic value. It arises at the southern boundary of a large sandstone plate (Pantano Formation, lower Miocene). The aquifer permeability is given by pervasive fracturing related to high-angle normal faults. A ranking methodology proposed by Gargini et al. (2008) for Apenninic springs was applied to the spring. In the ranking method, S (Slope) type and T (Trans-watershed) type springs are differentiated according to differential elevation of the spring above the local base level, recession coefficient α and average base flow discharge. Labante is classified as T-type spring and therefore has the potential to sustain drinking water supply for local communities and to host freshwater habitats (Bertrand et al., 2012). A

Table 3

Worldwide distribution of recognized LPS used to validate the conceptual model. The characteristic type-cases, geographic location, bibliographic reference, and host lithologies are indicated. Backup reviews and generic compilations: Ford and Pedley (1996); Pedley (2009); Pentecost (2005); Jones and Renaut (2010); Bense et al. (2013); Capezzuoli et al. (2014); Riding (2000).

Conceptual LPS type	Location	Lithology	References
4	Mt. Carameto, Mt. Pelpi, Mt. Caio, Mt. Carpegna (northern Apennines, Italy)	Interbedded calcarenites and calcareous mudstones	Chelli et al. (2013), Carlini et al. (2015)
6	Val Pessola (northern Apennines, Italy)	Coarse- to medium-grained sandstones	KO pers. Comm.
3	Northern Calcareous Alps	Interbedded sandstones and shales	Linhart and Schagerl (2015)
1, 2, 6	East African Rift System	Interbedded sandstones and shales	Ashley et al. (2014)
1, 2, 6	Colorado Plateau (SE Utah, USA)	Interbedded sandstones–mudstones	Gratier et al. (2012), Frery et al. (2015), Prieswisch et al. (2014); Ricketts et al. (2014)
1, 2, 3	Western Alps (Central Switzerland)	Different lithologies	Wehrli et al. (2010)
5, 6	SE Utah (USA)	Interbedded sandstones–mudstones	Dockrill and Shipton (2010), Ogata et al. (2014)
5, 6	New Mexico (USA)	Interbedded sandstones–mudstones	Crumpler (2003)
6	East-central Utah	Interbedded sandstones–mudstones	Jung et al. (2014)
1, 2, 6	Eastern Tunisia	Different lithologies	Essefi et al. (2014)
4	Sierra del Montsec (Pyrenees, Spain)	Biohermal limestones	Rosell and Linares (2001)
1, 2, 6	Itaboraí basin (Southeastern Brazil)	Basement rocks with sedimentary cover	Sant'Anna et al. (2004)
1, 2, 3, 4, 6	Eastern Alps (Europe)	Carbonate and calciclastic deposits	Sanders et al. (2010a, 2010b), Sanders and Rott (2009), Sanders et al. (2011)
6	Labante (northern apennines, Italy)	Medium- to fine-grained calcarenites	Gargini et al. (2012)
2, 6	South Tibet	Metamorphosed carbonate and siliciclastic, and crystalline rocks	Zentmyer et al. (2008)
2, 4, 6	Southern Apennines (Italy)	Carbonate and calciclastic deposits	Ascione et al. (2013)
1, 2, 6	Kerch Peninsula (Crimea, Russia/Ukraine)	Layered limestones	Kokh et al. (2015)
1, 2	Fossil Creek (Arizona, USA)	Mainly aeolian sandstones	Schleicher (2011)
1, 2	Dalhousie Springs (Australia)	Mainly siliciclastic deposits	Clarke and Bourke (2011)
5, 6	Poland, central and western Carpathians	Biohermal limestones	Mastella and Rybak-Ostrowska (2012)
	South Australia	Different lithologies	Keppel et al. (2011)
2, 5, 6	Oman	Ophiolites	Olsson et al. (2014)
2, 5, 6	Sierra de Alfacara (Granada, southern Spain)	Calcarenites, limestones, gypsum, marls, sandstones and claystones	Martín-Algarra et al. (2003), Andreo et al. (1999), Prado-Pérez et al. (2013)
2, 5, 6	Northern Australia, Barkly karst	Limestones and dolomites	Carthew et al. (2006)
1, 2, 6	Expedition Fjord, Canadian High Arctic	Gypsum-anhydrite	Omelson et al. (2000)
1, 2	Santa Barbara, California (USA)	Limestones and calcareous mudstones	Ibarra et al. (2014)
5, 6	Longmenshan, southwestern China	Mixed with dominant carbonates	Shi et al. (2014)
2, 5, 6	Plitvice, Croatia	Limestones and dolomites	Golubić et al. (2008)
2, 6	Slieve Bloom, Ireland	Limestones and dolomites	Heery (2007)
1, 2, 6	High Andes, Argentina	Conglomerates, sandstones, and intraclastic limestones	Valero-Garcés et al. (2001)
1, 2, 6	Taung, South Africa	Limestones and clastic-soil cover	McKee (2010)
1, 2, 6	Sears Lake, California (USA)	Conglomerates, sandstones, and intraclastic limestones	Guo and Chafetz (2012)
1, 2, 6	High Andes, Peru	Different lithologies	Acosta and Prat (2011)
2, 5, 6	Paris Basin, France	Limestones and clastic-soil cover	Freytet and Plet (1996)
1, 2, 6	Pyramid Lake, Nevada	Unlithified lithologies	Benson (2004)
1, 3	Guadalajara, Spain	Unlithified lithologies	Pedley et al. (2003)

recharge area of 2.61 km² was determined for the spring via numerical modelling (Gargini et al., 2014; Piccinini et al., 2014).

Carbonate-rich groundwater arising from the spring allows the deposition of SAL, which occurs in correspondence of a morphologic drop able to produce a spring-waterfall. SAL deposits grow at the waterfall front with the formation of a prograding flat surface. An important system of primary caves (the biggest in Italy) is hosted in these deposits. The caves and the waterfall represent the main touristic attractions of the site, while at least three different habitats hosted in the SAL (following the EU Habitat directive; EU-HD, 1992) are main grounds for its naturalistic and environmental interest as a Groundwater Dependent Ecosystem (GDE).

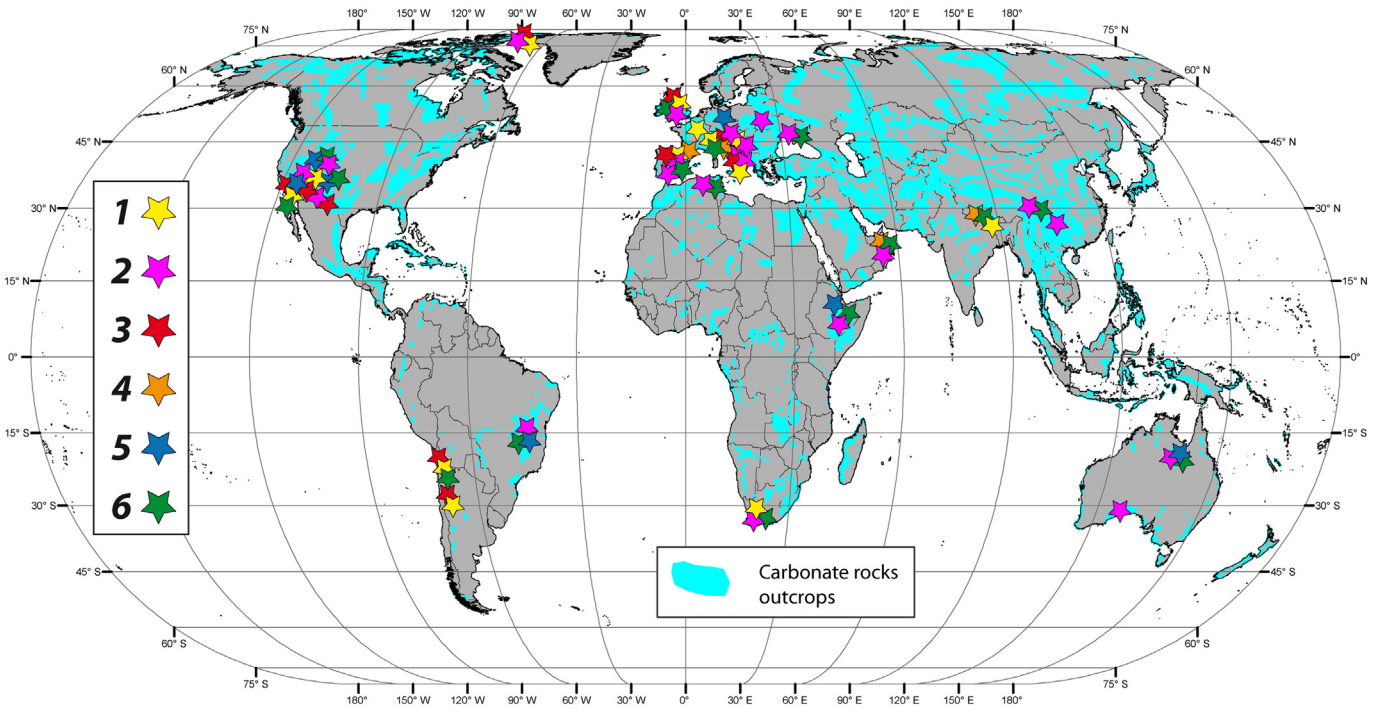
The Labante site (i.e. caves and waterfall) has been affected by human activities since the ancient age (e.g., exploitation of SAL as building materials in the Etruscan age, cave frequentation). Until a decade ago the site was in a general state of abandon and decadence, due to lack of guidelines and regulations for its preservation. Since 1993 the spring is exploited to gain drinking water. The water diversion occurs up-gradient from the SAL deposits (Fig. 5), causing a significant decrease of flow rate at the waterfall (up to 85% during the dry season). This produced a significant decrease in the progradation rate of the SAL front during the last two decades (Piccinini et al., 2014).

Despite the significant impact of water diversion, a biological survey based on sampling performed in 2011 revealed the occurrence of one of the most characteristic SAL species: the microalga *Oocardium stratum*

(ER, unpublished data). Moreover, the typical LPS mosses *Palustriella commutatata* and *Eucladium verticillatum* were found to be dominant among bryophytes (Daniel Spitale, unpublished data). Among the diatom microalgae several species characteristic of LPS springs including hydropetric microhabitats were found (e.g., *Gomphonema lateripunctatum*, *Delicata minuta*, *Denticula elegans*, *Cymbella diminuta*). This characteristic biota is not concentrated in the waterfall (where more widespread rheophilic species prevail) but distributed on the wet flanks of the large SAL complex. To ensure the protection of this biota and prevent deterioration, it is thus necessary to change the runoff pattern, now somewhat channelled towards the waterfall for scenic reasons (Fig. 5), into a more diffuse seepage on wide surfaces of the SAL complex. A minimum sustainable discharge towards the SAL complex should be warranted and estimated by means of tailored biological and geological (SAL deposition rates etc.) investigations.

In order to restore conflicts between human exploitation and GDE protection several measures have been developed and adopted since the last fifteen years: 1) the Labante site was included in several networks at the regional, national, and international scale (e.g., the European “Natura 2000” network, as “Site of Community Importance”); 2) hydrogeological researches were performed to identify the spring recharge area and to define a physically-based protection area (Piccinini et al., 2014); 3) public meetings were organized with the stakeholders (local authorities and citizens) in order to raise awareness on the naturalistic and environmental significance of the site; 4) guidelines were

Worldwide LPS distribution by types
(case studies used to validate the model)



Source: University of Auckland (New Zealand) - http://web.env.auckland.ac.nz/our_research/karst/

Fig. 3. Map showing the worldwide distribution of LPS types recognized in the available literature.

compiled to address the management of the drinking-water exploitation (minimum sustainable flow) and the site conservation (sustainable tourism, e.g.: proper access ways, monitoring of human activities potentially negative for the GDE).

7. Conclusions

Springs and spring-associated limestones have been classified before with highly different perspectives (e.g., according to types of

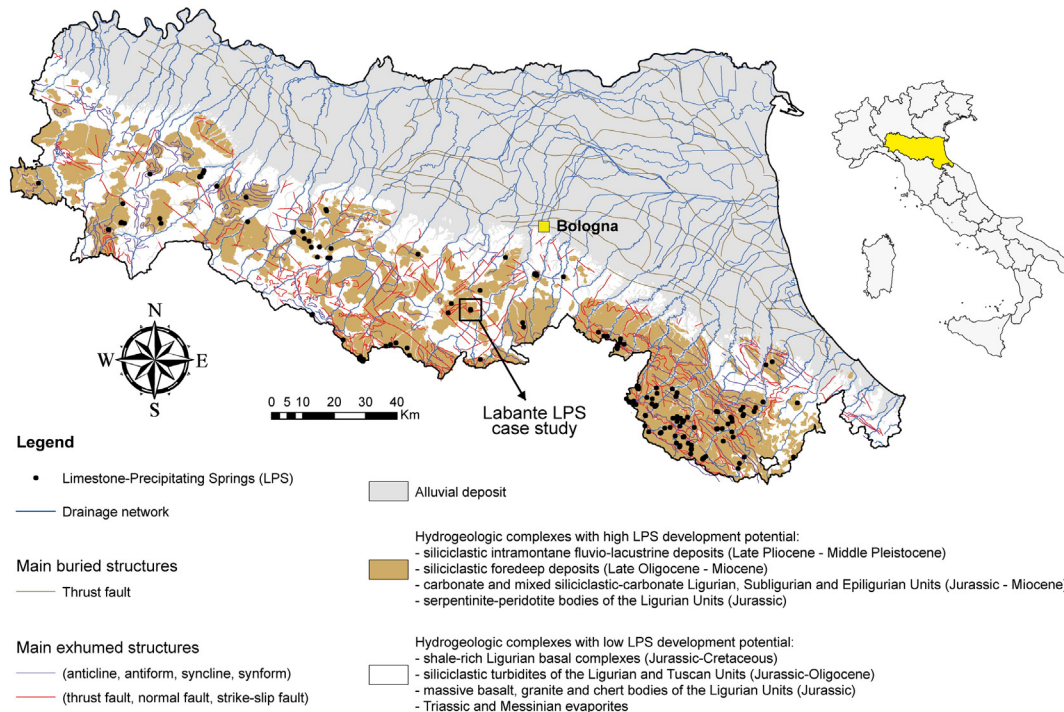


Fig. 4. Areas of high incidence of LPS occurrence as predicted by the conceptual model for the Emilia-Romagna Region. Recognized and ground-verified LPS are marked.

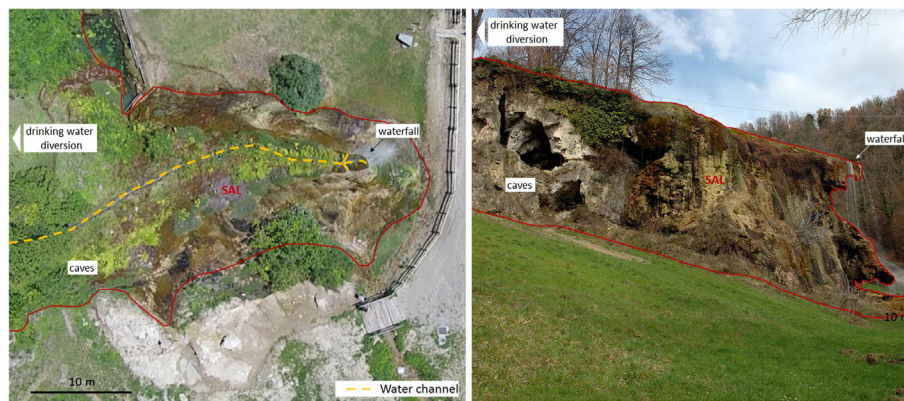


Fig. 5. View from above (left; summer; aerial view) and from the side (right; winter) of the Labante SAL. The SAL outcrop is outlined in both views (red line). The man-made channelization of water towards the waterfall is highlighted on the aerial view (yellow line). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

sedimentary facies, geomorphology, or botany) (see also review in Ford and Pedley, 1996). Our classification presented herein emphasizes the combined structural–geomorphological–hydrogeological aspect of LPS, so it seems best suited for regions of differentiated relief, such as highlands and mountain ranges. Our concept is not explicit on the type, facies architecture or size of spring-limestone deposystems; these features can be categorized in classification schemes presented previously (e.g., Pedley, 1990; Ford and Pedley, 1996; Carthew et al., 2003; Capezzuoli and Gandin, 2004; Jones and Renaut, 2010; Keppel et al., 2011). In mountain ranges, because of topographic relief and steep slopes, by far most spring limestone deposystems may pertain to the ‘perched springline (slope system)’ type of Ford and Pedley (1996, p. 123 f.). To achieve a more distinguishing characterization, depending on area, it may be desirable to further divide the perched-springline type into subgroups (e.g., Sanders et al., 2010a, 2010b). Travertines or hot-spring limestones commonly are associated with volcanism and/or with active faults guiding fluid ascend; therefore, travertines can be recorders of neotectonic deformation (e.g., Hancock et al., 1999; Minissale et al., 2002; Capezzuoli and Sandrelli, 2006; Zentmyer et al., 2008; De Filippis and Billi, 2012). With respect to the association of travertines with faults, mainly the structural aspects of our concept also are applicable to hot LPS.

Herein, Limestone Precipitating Springs (LPS) are understood as springs of ambient-temperature waters (no geothermal contribution) that achieve sufficient oversaturation for CaCO_3 -mainly by physical CO_2 -degassing and photosynthetic activity- to deposit limestone.

LPS support specific calcifying organisms, mainly cyanobacteria, algae, and mosses. The invertebrate fauna is of low diversity, and is limited to a very few adapted specialists coexisting with several generalists tolerating the permanent environmental stress by carbonate precipitation.

LPS are found on all continents but do not have a special protection status in most countries yet. This contrasts with the current situation in Europe where LPS are the most protected spring type (listed in Annex I of the Habitat Directive). Special protection status is primarily due to aesthetic, cultural and touristic reasons, and only secondarily to the scientific interest as key sites for ongoing geogenic processes including bioacalcification.

Table 4

Hydrodynamic, physical, chemical, and topographic features of the Labante spring. Data obtained from a continuous monitoring performed at spring 2003 and in 2010–2011.

	min	max	mean
Altitude (m a.s.l.)		612	
Q (L s^{-1})	3	28	13
T ($^{\circ}\text{C}$)	8.5	12.9	10.8
Conductivity 25 $^{\circ}\text{C}$ ($\mu\text{S cm}^{-1}$)	623	639	626
pH	7.4	8.0	7.7

To support mapping of LPS in fulfilment of the Habitat Directive, we developed a conceptual model based on fundamental stratigraphic and structural conditions to predict where LPS are more likely to occur in a particular region, with a focus on the geologic structure. This should facilitate an integrated view on spring phenomena and help optimizing management. The main impacts on LPS are due to inappropriate management underlain by missing awareness. It is thus important to disseminate knowledge on spring habitats, and to urge the application of flow splitters to sustain long-term persistence of key biota.

In Europe, a focus on LPS has the potential to flagship a more widespread and effective conservation of springs in general, since natural springs other than LPS harbour even much higher diversity of biota.

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